# College of Engineering Mechanical Engineering Department 

## Second Stage

## THERMODYNAMICI

## CHAPTER THREE - THE WORKING FLUID

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## Lecture no. 1 - How to use steam tables

## Case \#1 : Saturated Steam :

Required One main property like pressure or temperature with steam saturated condition (Dry) :
Example 1 : Find $u, h$, and $v$ for saturated steam at a pressure of 10 bar ?

## Solution :

Ask ? What the type of the steam ?
Look to (Saturated) which mean the steam is Dry and the Dryness fraction is 1 , therefore the following can apply :
$\mathrm{u}=\mathrm{ug}, \mathrm{h}=\mathrm{hg}$, and $\mathrm{v}=\mathrm{vg}$ at steam pressure 10 bar

## Lecture no. 1 - How to use steam tables-Cont.



## Lecture no. 1 - How to use steam tables-Cont.

## Case \#2 : Wet Steam :

Required two main properties like : pressure and temperature or pressure with dryness fraction Example 2 : Find $u, h$, and $v$ for wet steam at 10 bar with dryness fraction ( $x=0.85$ ) ?

## Solution :

Ask ? What type of the steam ?
Look to (wet) which mean the steam is wet and the Dryness fraction is less than 1 , therefore the following can apply :

$$
\begin{gathered}
u=u f+x(u g-u f), \\
h=h f+x h f g \text { or } h=h f+x(h g-h f) \\
\text { and } v=x v g
\end{gathered}
$$

## Lecture no. 1 - How to use steam tables-Cont.



## Lecture no. 1 - How to use steam tables-Cont.

## Case \#3 : Superheated Steam :

Required two main properties like pressure and temperature
Example 3 : Find $u, h$, and $v$ for steam at a pressure of 10 bar and a temperature of $200^{\circ} \mathrm{C}$ ?
Solution :
Ask ? What type of the steam ? Here we need to make simple check:
Go to table 8 , page 13 , and look to the temperature at a pressure of 10 bar?
CHECK : $\boldsymbol{t}_{\text {sat }}=179 . \mathbf{9}^{\circ} \mathrm{C}$ ( Saturation temperature)
The given $t=200^{\circ} \mathrm{C}$ which is greater than tsat. ( $\mathrm{t}>\mathrm{t}$ sat. ) ,
The steam is Superheated

## Lecturer no. 1 - How to use steam tables-Cont.

Solution
To find enthalpy , $h$ use table 9 page 16
$\mathrm{h}=2827 \mathrm{~kJ} / \mathrm{kg}$

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## Lecture no. 1 - How to use steam tables-Cont.

Solution cont. :
To find specific volume ,
$\checkmark$ use table 11 page 18

Density , $\rho=4.86 \mathrm{~kg} / \mathrm{m} 3$
$v=1 / \rho=0.2057 \mathrm{~m} 3 / \mathrm{kg}$

## Lecture no. 1 - How to use steam tables-Cont.

Solution cont. :

## To find internal energy, u use table 12 page 19

$\mathrm{u}=2621 \mathrm{~kJ} / \mathrm{kg}$


## Lecture no. 1 - How to use steam tables-Cont.

## HOME WORK :

Q1 : Calculate the dryness fraction, internal energy, and specific volume for steam at 8 bar and the specific enthalpy of $2650 \mathrm{~kJ} / \mathrm{kg}$.

Q2 : Steam at a pressure of 100 bar and specific volume of $0.02242 \mathrm{~m} 3 / \mathrm{kg}$, calculate its temperature, specific enthalpy and internal energy.

Q3: Steam at 150 bar and specific enthalpy of $2979 \mathrm{KJ} / \mathrm{kg}$, determine the temperature, specific volume and the internal energy.

Q4: A vessel of a volume 0.03 m 3 contains dry saturated steam at 20 bar, calculate the mass of steam in vessel and the enthalpy of this mass.

## Lecture no. 1 - How to use steam tables-Cont.

Steam at 7 bar and $250^{\circ} \mathrm{C}$ enters a pipeline and flows along it at constant pressure. If the steam rejects heat steadily to the surroundings, at what temperature will droplets of water begin to form in the vapour? Using the steady-flow energy equation, and neglecting changes in velocity of the steam, calculate the heat rejected per kilogram of steam flowing.

$$
\left(165^{\circ} \mathrm{C} ; 191 \mathrm{~kJ} / \mathrm{kg}\right)
$$

0.05 kg of steam at 15 bar is contained in a rigid vessel of volume $0.0076 \mathrm{~m}^{3}$. What is the temperature of the steam? If the vessel is cooled, at what temperature will the steam be just dry saturated? Cooling is continued until the pressure in the vessel is 11 bar; calculate the final dryness fraction of the steam, and the heat rejected between the initial and the final states.

$$
\left(250^{\circ} \mathrm{C} ; 191.4^{\circ} \mathrm{C} ; 0.857 ; 18.5 \mathrm{~kJ}\right)
$$

