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Design, synthesis, and biological evaluation of some new charge transfer complexes as a combination model

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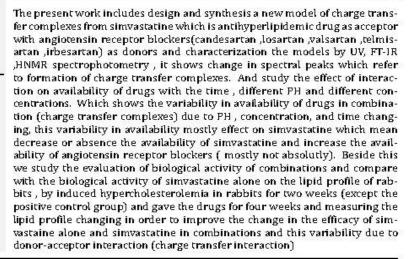
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ABSTRACT



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INTRODUCTION

The term charge transfer complex (CTC) was first used by Mulliken. He describes a new type of adduct to describe the behavior of certain molecular groups, which do not adhere to the traditional patterns of ionic, covalent, and hydrogen bonding components. While these adducts generally retain

some of the component properties, some changes are obvious For example, its solubility, its diamagnetic and paramagnetic susceptibility. Charging interactions within a molecular complex consisting of an electron donor D and an electron acceptor A include resonance with charging transfer from D to A (Abdulredha, 2015).

It is important phenomenon in the process of biochemical and bioelectrochemical energy. The term charge transfer gives kind of complex resulting from donor and acceptor interactions with the formation of weak bonds and widely discussed by Foster. In molecular interactions between electron donors and receivers are correlated with formation of strongly colored charge Transfer complexes (CTCs) which absorb radiation in visible region. Important processes in biological systems are molecular complexation and structural recognition. Drug action, catalysis of enzymes and movement of ions via lipophilic membranes all require complexation.