#### **ORIGINAL PAPER**



# **Characterization of** *Rubus fruticosus* **L. berries growing wild in Morocco: phytochemical screening, antioxidant activity and chromatography analysis**

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#### **Abstract**

*Rubus fruticosus* L. is a widespread shrub species which has recently attracted a great attention due a plethora of diferent pharmacological activities. A comparative study of the volatile profle and polyphenols along with the antioxidant activity of 27 blackberries collected in 3 diferent locations of the Tangier–Tetouan–Al Hoceima region in Northern Morocco, between June and August 2018, is reported. In terms of antioxidant activity, the highest  $IC_{50}$  values were attained for the *Rubus fruticosus* EtOAc extract coming from Beni Messaouar (2.5 mg/mL±0.01). Concerning the volatile content of the *Rubus fruticosus n*-hexane extract belonging to Beni Messaouar, a total of 42 compounds were detected and oleic acid turned out to be the most abundant one (14.49%), whereas among the 29 polyphenols detected in the *Rubus fruticosus* EtOAc extract, coming from the same location, quercetin-3-O-glucoside occurred in major concentration (364.58 mg/kg). This is the frst report on the physico-chemical and phytochemical properties of Moroccan *Rubus fruticosus* highlighting how these fruits do have a great potential as natural source of antioxidant compounds to be used as a nutraceuticals or functional foods.

**Keywords** *Rubus fruticosus* L. · Blackberry · Physico-chemical analyses · Antioxidant activity · Volatiles · Polyphenols · Liquid chromatography

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## **Introduction**

Blackberries (*Rubus fruticosus*) are wild fruits that have a delicious taste, pleasant favor, good appearance and an excellent nutritional profle. These fruits are eaten fresh or processed to make food products such as jam, tea, ice cream, desserts, jellies, and baked goods. The pigments extracted from these berries are employed in many application felds [[1](#page-9-0)].

*Rubus fruticosus* has a particular importance due to its high nutrient content. Its importance has become greater with the concept of "functional food" which made this fruit more popular [\[2\]](#page-9-1). This berry has various health benefts and since ancient time, this species has been recognized for their use in folk medicine: in particular due to its anti-infammatory properties, it can be advantageously employed for the cure of gums, ulcers of the oral cavity and cough [[3\]](#page-9-2). It is also endowed with anti-diarrheal, diuretic and anti-hemorrhoidal properties [[4–](#page-9-3)[6](#page-9-4)]. Moreover, it has an anxiolytic power by depression of the central nervous system [[7](#page-9-5)]. In addition to these activities, a protective effect against cognitive difficulties has been demonstrated [[8](#page-9-6)]. *R. fruticosus* berries are notable for their high nutritional contents of vitamin C, vitamin B, dietary fber, α-tocopherol, tocotrienol, calcium, potassium, magnesium, carotenoids, linoleic acid and linolenic acid [\[7](#page-9-5), [9](#page-9-7)]. Bioactive compounds include phenolic compounds such as ellagic acid and anthocyanins [[10–](#page-9-8)[12](#page-9-9)]. Furthermore, results obtained by GC–MS showed the presence of six major compounds, namely hexadecanoic acid methyl, 9,12-octadecadienoic acid methyl ester, 9,12,15-octadecatrienoic acid methyl ester, phytol, phthalic acid diisooctyl ester and vitamin E, occurring in the leaf extract, whereas four major compounds, namely 2-furancarboxaldehyde, hexadecanoic acid methyl ester, 12-octadecanoic acid methyl ester and phthalic acid diisooctyl ester, were found in each of the stems and the roots extracts [\[13](#page-9-10)]. Other results proved that the contents of sugars, total phenolic, total favonoid, and anthocyanins increase as the fruit development is advanced (ripening) [\[14\]](#page-10-0).

Considering the high-water content, these fruits do present an irrelevant presence of macromolecules, e.g., proteins lower than 3.5% whereas lipids up to 1%; on the other hand, among minerals, potassium was reported to be the most abundant one  $[15-17]$  $[15-17]$  $[15-17]$  $[15-17]$ .

The aim of this study was to evaluate the physicochemical properties and well as the antioxidant activity of 27 blackberries collected from 3 diferent locations of the Tangier–Tetouan–Al Hoceima region in Northern Morocco, between June and August 2018. The phytochemical profle of the *R. fruticosus* extracts namely ethylacetate (EtOAc) and MeOH:water (80:20 *v/v*), coming from Beni Messaouar (BM), was carried out by HPLC coupled to PDA and MS detection to determine the polyphenolic content. Besides, the *n*-hexane fraction, belonging to the same location, was evaluated in terms of volatile content, investigated by GC–MS.

# **Materials and methods**

#### **Chemicals**

Folin–Ciocalteu phenol reagent was obtained from Fluka. Reference materials (gallic acid, *p*-hydroxybenzoic acid, *p*-coumaric acid, cafeic acid, vanillin, rutin, kaempferol, quercetin and cyanidin-3-*O*-glucoside) were obtained from Merck Life Science (Merck KGaA, Darmstadt, Germany). 2,2-Diphenyl-1-picrylhydrazyl (DPPH) and butylated hydroxytoluene (BHT) were purchased from Sigma (St. Lois, MO). LC–MS grade methanol, acetonitrile, acetic acid, EtOAc, and water were purchased from Merck Life Science (Merck KGaA, Darmstadt, Germany). All other chemicals were attained from Sigma (St. Louis, MO).

#### **Samples and sample preparation**

Twenty-seven wild fruits (*R. fruticosus*) were harvested in Tangier–Tetouan–Al Hoceima region in the extreme northwest of Morocco. Specifcally, three fruits, in three diferent months, viz*.* June, July and August 2018 were collected in the following locations: Had Gharbia (HG), Ghezaoua (GH), and BM. Specifcally, the harvest areas were between the longitudes 5° 55′ 50.7"; 5° 31′ 34"; and 5° 43′ 31.3" and the latitudes 35° 30′ 54.3"; 34° 55′ 43"; and 35° 28′ 41.3" and, respectively, for HG, GH, and BM, respectively. After collection, fruits were stored at  $-10$  °C at the Laboratory of Valorization of Resources and Chemical Engineering, Abdelmalek Essaadi University, Tangier, Morocco.

5 g of lyophilized powder underwent a defatting step by adding three times 50 mL of *n*-hexane; afterwards, it was dried and homogenized with 50 mL of two solvents with increased polarity, namely EtOAc and MeOH:water (80:20 *v/v*). Each fraction was extracted using an ultrasound bath (130 kHz) for 45 min. After centrifugation at 5000 g for 5 min, the supernatant fltered through a paper flter, dried, reconstituted with MeOH:water, 80:20 (*v/v*), and then, fltered through 0.45 μm Acrodisc nylon membrane (Merck Life Science, Merck KGaA, Darmstadt, Germany) prior to HPLC–PDA–ESI/MS analysis [[18\]](#page-10-3).

#### **Physico‑chemical analyses**

Physico-chemical determinations were carried out according to the AOAC International Standard Methods. Parameters

detected were pH, refractive index (RI), total soluble solids (TSS), ratio sugar/acidity (S/A), dry matter content (DM %), ash (%), total sugars (TS%), reduce sugars (RS%), lipid content (mg/g), protein content (mg/g) and vitamin C content (mg/g).

## **Phytochemical screening**

Phytochemical screening was performed according to a previously published work [[18](#page-10-3)].

## **Test for starch**

A small portion of the extract is treated with boiling water for 30 s; afterwards, it is mixed with boiled ethanol for a few minutes. The addition of iodine provides a blue–black color.

## **Test for saponosides**

Roughly, 2 g of the powdered sample is boiled in 20 mL of distilled water in a water bath and fltered. 10 mL of the fltrate is mixed with 5 mL of distilled water and shaken vigorously for a stable persistent froth. The frothing is mixed with 3 drops of olive oil and shaken vigorously, leading to the formation of an emulsion.

## **Test for favonoids and anthocyanins**

The extract is treated with concentrated sulphuric acid. Appearance of yellowish orange shows the presence of anthocyanins, yellow to orange color shows the presence of favones, and orange to crimson shows the presence of favanones.

## **Test for tannins**

Roughly, 0.5 g of the dried powdered sample is boiled in 20 mL of water in a test tube, and then, fltered. A few drops of 0.1% ferric chloride is added and observed for brownishgreen or a blue–black coloration.

# **Test for sterols and steroids**

1 mL of concentrated sulphuric acid is added to 1 g of plant extract and allowed to stand for 5 min. After shaking, the formation of golden yellow color in the lower layer indicates the presence of sterols and steroids.

# **Test for mucilages**

Roughly, 5 mg of the extract is dissolved in a vial containing 0.5 mL of deionized water. The vial is heated in a laboratory oven at 100 °C for 5 min and allowed to cool. The liquid is then centrifuged for 1 min and three drops of the supernatant are mixed with 0.5 mL of *o*-toluidine solution. The solution is then heated in an oven for 10 min at 100 °C. Variation of color indicates the presence of mucilages.

## **Test for coumarins**

Roughly, 5 mg of the extract is treated with a solution of dimethylamino-benzaldehyde (5% ethanol), and then, acidifed by bubbling gaseous hydrochloric acid. The orange color indicates a positive test.

## **Test for alkaloids**

A small portion of the extract is stirred separately with 1 mL of diluted hydrochloric acid and fltered. The fltrate is treated with Dragendroff's reagent. Appearance of organic precipitate shows the presence of alkaloids.

## **Test for anthraquinones**

A small portion of the extract is boiled with dilute sulphuric acid. Filtered and cooled. The fltrate is extracted with chloroform or benzene and dilute ammonia is added to it. The ammonical layer becomes pink to red due to the presence of anthraquinones.

## **Test for iridoids**

5 g of each extract is mixed in 2 mL of chloroform and concentrated  $H_2SO_4$  (3 mL) are carefully added to form a layer. A reddish-brown coloration of the interface shows positive results for the presence of iridoids.

## **Test for glycosides**

Small quantity of the extract is hydrolyzed with 5 mL of hydrochloric acid for few hours on a water bath and the hydrolysate was subjected to Fehling's test. To 2 mL of Fehling's solution (1 mL of Fehling's A and 1 mL of Fehling's B solution), 2 mL of the extract is added, mixed and boiled. Appearance of yellow or red color precipitate indicates the presence of reducing sugars.

# **GC–MS analyses**

GC analyses of the *n*-hexane fraction of the nine *R. fruticosus* samples, coming from BM, were performed on a GC–MS-QP2020 system (Shimadzu, Kyoto, Japan) equipped with an AOC-20i system auto-injector.

Separations were performed on an SLB-5 ms column (30 m × 0.25 mm I.D.; 0.25 µm; Merck Life Science, Merck KGaA, Darmstadt, Germany). The initial temperature was set at 50 °C, afterwards increased up to 350 °C (increase rate: 3 °C/min; holding time: 5 min).

GC–MS parameters were as follows: injection temperature, 280 °C; injection volume, 1.0  $\mu$ L (split ratio: 10:1); pure helium gas (99.9%); linear velocity, 30.0 cm/s; inlet pressure, 26.7 kPa; full scan mode mass range, 40–660 m*/z*; event time, 0.2 s; EI source temperature, 220 °C; interface temperature, 250 °C. Relative quantity of the chemical compounds present in each sample was expressed as percentage based on peak area produced in the GC chromatogram.

Compounds were identifed using the FFNSC 4.0 (Shimadzu Europa GmbH, Duisburg, Germany), and "W11N17" (Wiley11-Nist17, Wiley, Hoboken, NJ, USA; Mass Finder 3). Each compound was identifed applying a MS similarity match and an LRI flter. Linear retention indices (LRI) were calculated using a C7–C40 saturated alkane reference mixture (49452-U, Merck Life Science, Merck KGaA, Darmstadt, Germany).

#### **HPLC–PDA/ESI–MS analyses**

HPLC analyses of the nine *R. fruticosus* samples (EtOAc and MeOH:water, 80:20 *v/v* extracts), were performed on a Shimadzu liquid chromatography system (Kyoto, Japan), consisting of a CBM-20A controller, two LC-30AD dualplunger parallel-flow pumps, a DGU-20A5R degasser, a CTO-20AC column oven, a SIL-30AC autosampler, an SPD-M30A photodiode array detector, and an LCMS-8050 triple quadrupole mass spectrometer, through an ESI source (Shimadzu, Kyoto, Japan).

Separations were carried out on  $150 \times 4.6$  mm; 2.7 µm Ascentis Express RP C18 column (Merck Life Science, Merck KGaA, Darmstadt, Germany). The mobile phase was composed of two solvents: water/acetic acid (99.85/0.15 *v/v*, solvent A) and acetonitrile/acetic acid (99.85/0.15 *v/v*, solvent B). The fow rate was set at 1 mL min−1 and a gradient elution program was followed: 0–5 min, 5% B, 5–15 min, 10% B, 15–30 min, 20% B, 30–60 min, 50% B, and 60 min, 100% B. PDA range was in 200–400 nm and monitored at  $\lambda$  = 280 nm (sampling frequency: 40.0 Hz, time constant: 0.08 s). MS conditions were as follows: scan range, *m/z* 100–800; scan speed, 2500 μ s<sup>-1</sup>; event time, 0.3 s; nebulizing gas (N<sub>2</sub>) flow rate: 1.5 L min<sup>-1</sup>; drying gas (N<sub>2</sub>) flow rate, 15 L min−1; interface temperature, 350 °C; heat block temperature, 300 °C; DL (desolvation line) temperature: 300 °C; DL voltage, 1 V; interface voltage: − 4.5 kV.

Calibration curves ( $R^2 \ge 0.997$ ) of nine polyphenolic standards were used for the quantifcation in sample extracts, namely gallic acid, *p*-hydroxybenzoic acid, p-coumaric acid, cafeic acid, vanillin, rutin, kaempferol, quercetin and cyanidin-3-*O*-glucoside. Five concentration levels were investigated in the range from 1 to 500 mg  $\text{Kg}^{-1}$ .

#### **Determination of the polyphenolic content**

Total phenol (TP) content for both EtOAc and MeOH:water  $(80:20 \nu/\nu)$  extracts was estimated using the Folin–Ciocalteu method [\[18](#page-10-3)] with some modifcations. Gallic acid was used as standard (10, 25, 50, 100, and 200 ppm) and total phenolic content was measured at 755 nm and was expressed as mg of gallic acid equivalents (GAE)/g dry mass (DM). Total favonoid (TFV) content for both EtOAc and MeOH:water (80:20 *v/v*) extracts was determined according to the method of Zhishen et al. [\[19](#page-10-4)]. A known volume of each extract was placed in a 10 mL of volumetric fask and flled with 5 mL of distilled water and  $0.3$  mL of NaNO<sub>2</sub> (1:20). Afterwards,  $3 \text{ mL of } AICl<sub>3</sub> (1:10)$  and  $2 \text{ mL of } NaOH (1 M)$  were added. The solution was mixed and the absorbance was measured against a blank at 510 nm. Results were expressed as mg of quercetin equivalents (QE)/g DM. Total anthocyanin (TA) content for both EtOAc and MeOH:water (80:20 *v/v*) extracts was estimated based on the differential  $pH (pH=1)$ and  $pH = 4.5$ ) with some modifications and expressed as mg of Pg-3-gluc/g DM [[20\]](#page-10-5). Total tannin (TT) content for both EtOAc and MeOH:water (80:20 *v/v*) extracts was determined by the vanillin method  $[21]$  $[21]$  $[21]$ ; briefly, 0.1–0.5 mL of extracts was taken and put into tubes covered with aluminum foil. Three milliliters of  $4\%$  vanillin (w/v) in methanol was added, and the tubes were shaken vigorously with a mixer. Afterwards, 1.5 mL of concentrated HC1 was pipetted and the tubes were shaken again. The absorbances were read at 500 nm after being allowed to stand for 20 min at room temperature and results expressed as mg of catechin equivalents  $(CE)/g$  DM.

#### **Determination of the antioxidant activity**

Free radical-scavenging DPPH. method for each sample was carried out following a slightly method described by Braca et al. [\[22](#page-10-7)]. A rapid TLC-screening method was frst performed using a 0.2% DPPH solution in MeOH. The spectrophotometric assay was carried out by adding 30 μL of a methanolic solution containing the single EtOAc and MeOH:water (80:20 *v/v*) extracts to 3 mL of a 0.004% MeOH solution of DPPH. Absorbance at 517 nm was determined after 30 min, and the percentage of activity was calculated. Butylated hydroxytoluene (BHT) was used as a positive control.  $IC_{50}$  was calculated from linear regression (%DPPH remaining radical versus sample concentration).

#### **Statistical analysis**

The experiments were carried out in triplicates and the results were expressed as the average of the three measurements  $\pm$  SD. The comparison of means between groups was performed with one-way analysis of variance (ANOVA) followed by Tukey test. Diferences were considered significant when  $p < 0.05$ .

## **Results and discussion**

*R. fruticosus* is a wide species present in the Mediterranean area. In total, 27 wild fruits were harvested in Tangier–Tetouan–Al Hoceima region. Three diferent locations, namely HG, GH and BM were selected and three fruits, in three diferent months, viz*.* June, July and August 2018 were collected.

#### **Determination of phenolic compounds**

The mean  $(n=9)$  physico-chemical parameters values obtained for the three diferent locations are reported in Table S1. The highest values of pH and ash% were found in *R. fruticosus* from the HG location; on the other hand, for TSS, S/A, and DM%, the most prominent values were attained for BM. Statistically, a signifcant diference was observed between the results of the three regions for all tested parameters  $(p < 0.05)$ . In terms of humidity, the values obtained for the three studied regions are lower than those reported by Hirsch et al. [\[23](#page-10-8)] for blackberry cultivars (*Rubus* spp.) distributed in the southern region of Brazil (from 84.8 to 90.3%); the same applies also to ash% values (from 0.27 to 0.49%) and pH values (from 2.8 to 3.1). The DM% contents of *R. fruticosus* are in agreement with another research where a percentage of 36.4% was attained [[24](#page-10-9)]. The S/A ratio obtained complied with an average

value obtained by other reports, viz*.* 3.1 [\[7\]](#page-9-5). It has been demonstrated that the variation in sugar content could be attributed to diferent factors, including stage of ripening, temperature, duration of sun exposure and weather conditions [[25\]](#page-10-10). In terms of TSS, studies on wild blackberries from diferent regions of Turkey and Serbia showed similar results, e.g., 11.3–13.1 and 12.0–15.6, respectively [[26,](#page-10-11) [27\]](#page-10-12). The latter also reported TS% and RS% values which are considerably lower than the ones reported in this study from 5.36 to 5.98 and from 1.32 to 1.46, respectively. The diferences in such results can be probably due to several factors such as climate and geographic distribution [[18](#page-10-3)]. Concerning the lipid and protein contents, the highest concentrations were obtained for the GH fruits (14.7 mg/g and 1.4 mg/g). Similar values of the same species were previously reported for proteins content, e.g., 1.39 mg/g [[7](#page-9-5)]. With regards to vitamin C, the highest value was attained for the berries of the HG locations  $(70.0 \pm 2.54 \text{ mg/g})$ which is three times over the one already found for the same species [\[7\]](#page-9-5).

## **Phytochemical screening**

The phytochemical screening of the 27 *R. fruticosus* fruits collected from the HG, GH and BM locations is reported in Table S2. The tested plant showed positive results for variable amounts of saponosides, flavonoids (flavones, isofavones), tannins, anthocyanins, sterols and steroids, coumarins, mucilages, alkaloids, iridoids and glycosides. Notably, sterols, steroids and mucilages were present in considerable amounts in the samples coming from the three



<span id="page-4-0"></span>**Fig. 1** GC–MS profle of the *n*-hexane fraction of one of the nine *R. fruticosus* samples, coming from BM. Only most abundant peaks are labeled

<span id="page-5-0"></span>**Table 1** List of compounds identifed in all the nine *R. fruticosus* samples from BM by GC–MS

No.	Compound	LRI (lib)	LRI (exp)	Similarity	Area %	Library
$\mathbf{1}$	$n$ -Nonanal	1107	1105	91	0.05	FFNSC <sub>4.0</sub>
$\overline{\mathbf{c}}$	(2E,4Z)-Decadienal 1295		1296	93	0.08	W11N17
3	(2E,4E)-Decadienal 1322		1320	93	0.08	FFNSC 4.0
4	Ethyl-decanoate	1399	1394	89	0.03	FFNSC 4.0
5	Vanillin	1394	1398	90	0.05	FFNSC 4.0
6	Methyl-dodecanoate 1527		1524	94	0.12	FFNSC 4.0
7	$n$ -Dodecanoic acid 1581		1563	95	0.16	FFNSC 4.0
8	Ethyl-dodecanoate 1598		1593	96	0.31	FFNSC 4.0
9	Methyl-tetrade- canoate	1727	1724	91	0.07	FFNSC 4.0
10	$n$ -Tetradecanoic acid 1773		1762	84	0.18	FFNSC 4.0
11	Ethyl-tetradecanoate1794		1793	95	0.18	FFNSC 4.0
12	Neophytadiene	1836	1836	93	0.12	FFNSC 4.0
13	Phytone	1841	1842	93	0.11	FFNSC <sub>4.0</sub>
14	Methyl-hexade- canoate	1925	1926	96	1.87	FFNSC 4.0
15	16-Hexadecanol- actone	1938	1942	84	0.14	FFNSC 4.0
16	$n$ -Hexadecanoic acid	1977	1965	95	4.03	FFNSC 4.0
17	Ethyl-palmitate	1993	1993	97	3.29	FFNSC <sub>4.0</sub>
18	Epimanool	2057	2060	87	0.05	FFNSC 4.0
19	Methyl-linoleate	2093	2093	92	5.04	FFNSC 4.0
20	Methyl-oleate	2098	2099	85	5.7	FFNSC 4.0
21	Phytol	2111	2111	91	0.41	FFNSC 4.0
$22\,$	Methyl-octade- canoate	2127	2127	92	0.31	FFNSC 4.0
23	Linoleic acid	2144	2138	95	8.95	FFNSC 4.0
24	Oleic acid	2142	2144	89	14.49	FFNSC 4.0
25	Ethyl-linoleate	2164	2161	94	10.89	FFNSC 4.0
26	Ethyl-stearate	2198	2194	95	0.6	FFNSC 4.0
$27\,$	(E)-Phytol acetate	2221	2213	91	0.06	FFNSC 4.0
28	$n$ -Tricosane	2300	2300	$90\,$	0.21	FFNSC 4.0
29	4.8.12.16-Tetra- methylheptadecan- 4-olide	2364	2349	95	0.33	W11N17
30	(9Z)-Octadecena- mide	2375	2362	94	1.82	W11N17
31	$n$ -Pentacosane	2500	2499	95	0.42	FFNSC 4.0
32	$n$ -Hexacosane	2600	2600	92	$0.07\,$	FFNSC 4.0
33	Tetracosanal	2632	2636	91	0.06	W11N17
34	$n$ -Heptacosane	2700	2700	95	1.1	FFNSC 4.0
35	$n$ -Octacosane	2800	2800	92	0.13	FFNSC 4.0
36	Squalene	2810	2813	$88\,$	0.19	FFNSC 4.0
37	$n$ -Nonacosane	2900	2900	90	1.11	FFNSC 4.0
38	δ-Tocopherol	2951	2947	94	3.22	W11N17
39	$\gamma$ -Tocopherol	3055	3054	96	8.88	W11N17
40	$n$ -Hentriacontane	3100	3100	95	0.29	FFNSC 4.0
41	Vitamin E	3138	3132	96	4.1	W11N17
42	$\gamma$ -Sitosterol	3351	3322	92	4.61	W11N17
	TOT. Identified				83.91	
	TOT. not identified				16.09	

<span id="page-6-0"></span>**Table 2** Polyphenols identifed in all the nine *R. fruticosus* EtOAc extracts from BM by HPLC–PDA/ESI–MS

Peak	Tentative identifica- $t_R$ (min) tion		Identification type	$\lambda_{\text{max}}$ (nm)	$[M-H]$ <sup>-</sup>	Fragments	Quantity (mg $kg^{-1}$
1	Gallic acid	6.26	PDA/MS	270	169	$\qquad \qquad -$	100.30
2	$3-p$ -Coumaroylquinic acid	8.86	PDA/MS	258-293	337		77.96
3	3-O-Caffeoylquinic acid	11.07	PDA/MS	280-321	353	179	39.56
4	$p$ -Hydroxybenzoic acid	12.28	PDA/MS	254	137	-	130.81
5	Unknown	13.90	PDA/MS	266	443	$265*$	
6	Vanillic acid	15.37	PDA/MS	$260 - 290$	167		147.24
7	3-Feruloylquinic acid 16.20		PDA/MS	321	367	$\overline{\phantom{0}}$	56.18
8	5-O-Caffeoylquinic acid	16.58	PDA/MS	321	353	179	70.71
9	Protocatechuic acid	19.71	PDA/MS	291	153	$\qquad \qquad -$	190.53
10	p-Coumaric acid	21.85	PDA/MS	305	163		23.02
11	Ferulic acid	24.82	PDA/MS	230-321	193		105.39
12	Vanillic acid deriva- tive	26.21	PDA/MS	281	363	167	61.13
13	Dihydroquercetin	27.07	PDA/MS	284	303	$\qquad \qquad -$	309.46
14	p-Coumaroyl tartaric 27.16 acid glucosidic ester		PDA/MS	268	475*	$\overline{\phantom{0}}$	45.32
15	Cyanidin-O-pentosyl 27.83 (hexoside)		PDA/MS	281-520	581	449	9.93
16	Unknown	28.15	PDA/MS	349	351	-	
17	Quercetin-3-O-glu- coside	28.50	PDA/MS	255-353	463	301	364.58
18	Ellagic acid hexoside 29.10		PDA/MS	352	463	301	
19	Quercetin-3-O-[6"- $O$ -(3-hydroxy-3- methylglutaryl) $-\beta$ -D-glucopyranoside]	30.96	PDA/MS	353	607	301	56.77
20	Cyanidin-O-glucoside 31.76		PDA/MS	284-520	449*	289*	7.07
21	Cyanidin-O-hexoside 32.20		PDA/MS	286-520	449*	289*	3.80
22	Cyanidin-O-pentoside 33.55		PDA/MS	354-520	449*	289*	6.70
23	Unknown	35.75	PDA/MS	281	359	329	
24	Ellagic acid pentoside 36.22		PDA/MS	267	433		-
$25\,$	Unknown	36.56	PDA/MS	261	549	263	
26	Luteolin	37.97	PDA/MS	287-356	285	$\overline{\phantom{0}}$	
27	Rutin	39.34	PDA/MS	361	609	301	41.61
28	Pelargonidin-succiny- 42.58 larabinoside or Pelargonidin-malo- nylrhamnoside		PDA/MS	$281 - 508$	503	271	1.27
29	Kaempferol	44.08	PDA/MS	291-359	285		6.33

\*Acquired in  $[M+H]$ <sup>+</sup> mode

Peak	Tentative identification $t_{\rm p}$ (min)		Identification type	$\lambda_{\text{max}}$ (nm)	mlz	Fragments	Quantity (mg/kg)
3	3-O-Caffeoylquinic acid 11.07		PDA/MS	280-321	353	179	30.86
12	Vanillic acid derivative 26.21		PDA/MS	281	363	167	32.90
14	<i>p</i> -Coumaroyl tartaric acid glucosidic ester	27.16	PDA/MS	268	$475+$	-	21.88
17	Ouercetin-3-O-glucoside $28.50$		PDA/MS	255–353	463	301	48.39
18	Ellagic acid hexoside	29.10	PDA/MS	352	463	301	—
19	Ouercetin-3- $O$ -[6"- $O$ -(3-hydroxy-3- methylglutaryl)- $\beta$ -D- glucopyranoside]	30.96	PDA/MS	353	607	301	8.96

<span id="page-7-0"></span>**Table 3** Polyphenols identifed in all the nine *R. fruticosus* MeOH/water extracts (80:20, *v/v*) from BM by HPLC–PDA/ESI–MS

regions. Notable amounts of catechic tannins occurred in the HG and GH location, whereas gallic tannins only in the HG one. Starch and anthraquinones were not found in any of the samples of *R. fruticosus* investigated. Such fndings are in agreement with previously published reports confrming that blackberries are a rich source of favonoids, tannins, sterols and steroids [\[7](#page-9-5), [28](#page-10-13)].

# **GC–MS analyses**

The GC–MS analysis of the *n*-hexane fraction of one of the nine *R. fruticosus* samples, coming from BM, is reported in Fig. [1](#page-4-0). A total of 42 compounds were detected (Table [1\)](#page-5-0) with a % of similarity ranging from 84 to 97%. Among them, lipids (oleic acid, 14.49%, ethyl-Linoleate, 10.89%, linoleic acid, 8.95%, and methyl-Linoleate, 5.04%), alkanes (n-Nonacosane, 1.11%, n-Heptacosane, 1.1%), tocopherols (γ-tocopherol, 8.88%, δ-tocopherol), sterols (γ-Sitosterol, 4.61%), aldehydes (*n*-Nonanal, 0.05%), etc. were positively identifed.

## **Determination of the polyphenolic content by HPLC–PDA/ESI–MS**

Qualitative and quantitative analyses of the *R. fruticosus* EtOAc and MeOH:water (80:20 *v/v*) extracts, coming from BM, were accomplished by HPLC–PDA/ESI–MS. A total of 29 and 6 compounds for EtOAc and MeOH/water (80:20,  $v/v$ ) extracts, respectively (Tables [2](#page-6-0) and [3](#page-7-0)), belonging to phenolic acids, anthocyanins and favonoids, were positively detected. As an example, Fig. [2](#page-7-1) shows the LC-PDA chromatogram of one of the nine *R. fruticosus* samples, coming from BM. Specifcally in the EtOAc extract, 14 phenolic acids belonging to hydroxybenzoic and cinnamic acids were found along with 5 anthocyanins and 6 favonoids, whereas 4 phenolic acids and 2 favonoids were identifed. Among them only rutin was reported as constituent of *R. fruticosus* from Croatia [\[10](#page-9-8)], whereas gallic acid, protocatechuic acid, vanillic acid and rutin were reported in *R. fruticosus* pomace from Bosnia and Herzegovina [[11\]](#page-9-11). From a quantitative point of view in the EtOAc extract, quercetin-3-*O*-glucoside

<span id="page-7-1"></span>**Fig. 2** HPLC–PDA  $(\lambda = 280 \text{ nm})$  fingerprint of the one of the nine *R. fruticosus* samples (EtOAc extract), coming from BM. For peak identifcation, see Table [2](#page-6-0)



<span id="page-8-0"></span>**Table 4** Total phenolic, flavonoid, anthocyanin, tannin contents and antioxidant activity IC<sub>50</sub> values of the 27 *R. fruticosus* fruits investigated  $(n=9)$ 



The results are expressed as means  $\pm$  SD (n=3). In each column, different letters mean significant differences (p <0.05)

<span id="page-8-1"></span>**Table 5** TSS, IR, TS, proteins, vitamin C, TPP, TFV, TA, TT and IC<sub>50</sub> values of the 27 *R. fruticosus* fruits investigated

R. fruticosus TSS		IR	TS	Proteins	Vitamin C TPP		<b>TFV</b>	TA	TT	$IC_{50}$	
<b>TSS</b>											
IR	0.999										
TS	0.233	0.189	1								
Proteins	0.537	0.549	0.040								
Vitamin C	0.151	0.106	0.968	$-0.197$							
TPP	$-0.627$	$-0.646$	0.137	$-0.887$	0.316						
<b>TFV</b>	0.276	0.232	0.996	0.038	0.971	0.102					
<b>TA</b>	0.232	0.189	0.974	0.129	0.929	0.001	0.975				
TT	0.227	0.182	0.796	$-0.114$	0.820	0.305	0.778	0.753			
$IC_{50}$	$-0.437$	$-0.433$	$-0.262$	0.256	0.302	0.106	$-0.321$	0.989	0.046		

turned out to be the most abundant one (364.58 mg/kg) followed by dihydroquercetin (309.46 mg/kg). Likewise Quercetin-3-O-glucoside was also the most abundant one in the MeOH/water (80:20, *v/v*) extract (48.39 mg/kg).

#### **Antioxidant activity and correlations**

It has been reported that the antioxidant activity of phenolic acids depends on the number and the position of hydroxyl groups related to the carboxyl functional group and it increases with hydroxylation degree [\[29\]](#page-10-14). Regarding *R. fruticosus* antioxidant activity, a research has shown an IC<sub>50</sub> equal to 2.14 mg/mL $\pm$ 0.12 [\[30\]](#page-10-15) and it has been demonstrated that antioxidant activity increases at increasing ripening stage with no signifcant diferences in free radical-scavenging activity between immaturity and semi-maturity for all samples [\[14](#page-10-0)]. Table [4](#page-8-0) reports total phenolic, total favonoid, total anthocyanin and total tannin contents

along with the antioxidant activity (IC $_{50}$  values) of the 27 *R. fruticosus* fruits investigated. The mean  $IC_{50}$  of each solvent fraction studied showed that the EtOAc extract showed the highest antioxidant power. Analysis by the ANOVA test showed a very signifcant diference between the results of the two fractions ( $p < 0.01$ ). Table [5](#page-8-1) reports the relationship between the  $IC_{50}$  values and TSS, IR, TS, proteins, vitamin C, TP, TFV, TA, TT and of the *R. fruticosus* fruits investigated. It was observed a strong correlation with  $R^2$  = 0.988 between  $IC_{50}$  and TA, while TT showed the weakest correlation  $(R^2=0.04)$  [[27](#page-10-12), [30–](#page-10-15)[32\]](#page-10-16). Several studies on berries and cherries have reported relationship between antioxidant activity and phenolic compounds and anthocyanin content [ $33-37$ ].  $R^2$  values of 0.97 were attained for TA and TFV as well as for TA and TS. Proteins were correlated negatively with phenolic compounds  $(R^2=0.89)$ , which was explained by their enzymatic interactions during ripening and/or fer-mentation [\[38,](#page-10-19) [39\]](#page-10-20).

It can be assumed that favonoid compounds, in particular quercetin-3-*O*-glucoside and dihydroquercetin, which were present at higher concentrations in the EtOAc extracts, contribute to some extent to the observed antioxidant efects. However, according to the results attained, the polyphenolic content (particularly favonoids) were not all consistent with the antioxidant activity. This could be due to the presence of other secondary metabolites responsible for this activity, which could have had potential synergistic or antagonistic effects. This latter could occur in the system based on additional components contained in this fruit, as well as interactions between phenolic compounds and plant matrix physical environment [\[40\]](#page-10-21). Some minerals (in particular iron) can complex with phenolic compounds infuencing their antioxidant activity [[41\]](#page-10-22). Since several characteristics and reaction mechanisms are likely involved, no single test accurately refect all antioxidants in a mixed or complex system [\[42](#page-10-23)]. The explanation about the absence of correlations between the different parameters and the antioxidant activity  $(IC_{50})$ , may be due to some parameters, e.g., solvent, concentration, structure, and size, that can afect the antioxidant capacity of extracts.

# **Conclusion**

Blackberries collected in three different regions of the Northern Morocco were evaluated. A comparative study of volatile and polyphenolic profles along with the antioxidant properties were evaluated and accomplished by both spectrophotometric and chromatographic approaches. No remarkable diferences were noticed for the three diferent regions investigated, except for the mean  $IC_{50}$  of each extraction solvent studied, which showed that the EtOAc extract does possess the highest antioxidant power. The evaluated physico-chemical and phytochemical parameters provided the evidence of an antioxidant potential, which explains some of the medicinal uses and pharmacological properties attributed to this species.

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supervision. LM: resources and project administration; FC: conceptualization, methodology, resources, writing—review and editing, and supervision.

## **Declarations**

**Conflict of interest** The authors declare that there is no confict of interest.

**Compliance with ethics requirements** This study does not contain any studies with human participants or animals performed by any of the authors.

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