

Electronic band structure and optoelectronic properties of SrCu₂X₂ (X = As, Sb): DFT calculation

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Received: 11 February 2014 / Accepted: 5 April 2014 / Published online: 23 April 2014
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Abstract All-electron-full potential linear-augmented plane wave method with Engel Vosko approximation was used for calculating the electronic structure, Fermi surface, and optical properties of SrCu₂X₂ (X = As, Sb). The calculated band structure and Fermi surface show that the metallic behavior of SrCu₂X₂ increases as one move from As to Sb. The calculated partial density of states shows that As-s/p/d, Cu-s/p, and Sr-s/p/d states are forming the Fermi surface for SrCu₂As₂, whereas Sb-s/p/d, Cu-s/p, and Sr-s/p/d states are forming the Fermi surface for SrCu₂Sb₂. The calculated densities of states at Fermi level and electronic specific heat are 14.2 (42.57) states/Ryd-cell and 2.60 (7.37) mJ/mol K² for SrCu₂As₂ (SrCu₂Sb₂). The complex optical dielectric function's dispersion and the related optical properties such as refractive indices, extension coefficient, absorption coefficient, reflectivity, energy loss function, and optical conductivity were calculated and discussed in detail. The optical properties show a considerable anisotropy between the two components.

Introduction

The investigation of novel high-temperature superconductors and its mechanism are always highlighted in condensed matter physics [1]. Among them, the AB₂M₂-type crystals are considered to have the largest family members of the crystals [2]. In AB₂M₂ type structure, A stands for alkali's, alkaline earth, or rare earth element, B represents transition metal, M symbolizes pnictogen or chalcogen. The ThCr₂Si₂ was the early discovered crystal in this group of compounds [3], as a consequence, this family of crystals is generally known as ThCr₂Si₂-type. These crystals got considerable attention because of its unusual characteristics such as superconductivity [4], intermediate valence [5], phase transition [6, 7], and different magnetic properties [8, 9].

The experimental results of these crystals established that by applying pressure or changing, the temperature induced first- and second-order phase transitions [10–14]. These phase transitions are related to the changes of the lattice parameters, in particular, the distance of M–M bonds between the layers of BM₄ tetrahedrons. The literature survey of AFe₂As₂ (A = Ba, Sr, Ca, Eu) visualized that they can be turned into superconductors either by applying pressure [15, 16] or by doping [17–19]. In the past years, much effort has been done on the experimental side for this class of compounds, whereas theoretical studies are relatively inadequate. In AB₂M₂, family Pfisterer and Nagorsen [20] for the first time synthesized SrCu₂As₂. Singh [21] performed the electronic structure calculation using the linearized augmented plane wave method within the local density approximation (LDA). The calculation exposed that the maximum density of states is covered by s and p states, the sp metal nature; the bands originated from Cu-d states nearly –3.0 eV (below Fermi level); therefore, SrCu₂As₂ was anticipated to be sp-band metals with Cu

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atoms contain a formal oxidation state of Cu^{+1} and chemically inert and nonmagnetic $3d^{10}$ state. But there was deficiency of other properties. Anand et al. [22], re-synthesized single crystals of Cu-based compounds BaCu_2Sb_2 , SrCu_2As_2 , SrCu_2Sb_2 and measure the structural, electronic, magnetic, and transport properties up to 300 K. They observed no phase transition in this range of temperature. Yan et al. [1] synthesized $\text{SrFe}_{2-x}\text{Cu}_x\text{As}_2$ single crystals and calculate the structural, magnetic, and electronic transport properties. In SrCu_2As_2 system, the sp-band metallicity with Cu in $3d^{10}$ electronic configuration is consequent to the valence state Cu^{1+} . As compared to SrCu_2As_2 , the almost unchanged Cu- $2p$ core line position in $\text{SrFe}_{2-x}\text{Cu}_x\text{As}_2$ indicates that partial Cu substitutions for Fe in SrFe_2As_2 may result in hole doping rather than the expected electron doping. No superconductivity is induced by Cu substitution on Fe sites, even though the structural/

spin density wave transition is gradually suppressed with increasing Cu doping [1].

Recently Lv et al. [23] calculated the thermal and elastic properties of SrCu_2As_2 by using CASTEP code [24] within generalized gradient approximation, adopted with Wu-Cohen (WC-GGA) [25] and norm-conserving pseudopotential.

Until now, no study in the literature has been found on optical properties for SrCu_2X_2 ($\text{X} = \text{As}, \text{Sb}$) compounds. Therefore, it has been thought of interest to study the optical properties of SrCu_2X_2 ($\text{X} = \text{As}, \text{Sb}$) compounds in addition to electronic structure. In this paper, full potential linear augmented plane wave (FPLAPW) method, have been used to calculate the density of states, electronic band structure, Fermi surface, and optical properties. The FPLAPW method has been proven to be one of the most accurate and flexible method, at reasonable computational expense [26, 27]. This method works with a true crystal

Fig. 1 Optimize unit cell of; **a** SrCu_2As_2 , **b** SrCu_2Sb_2

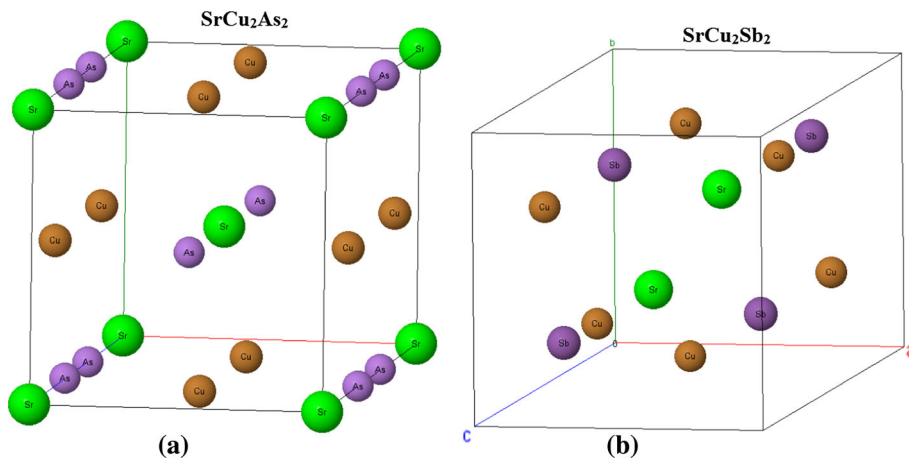
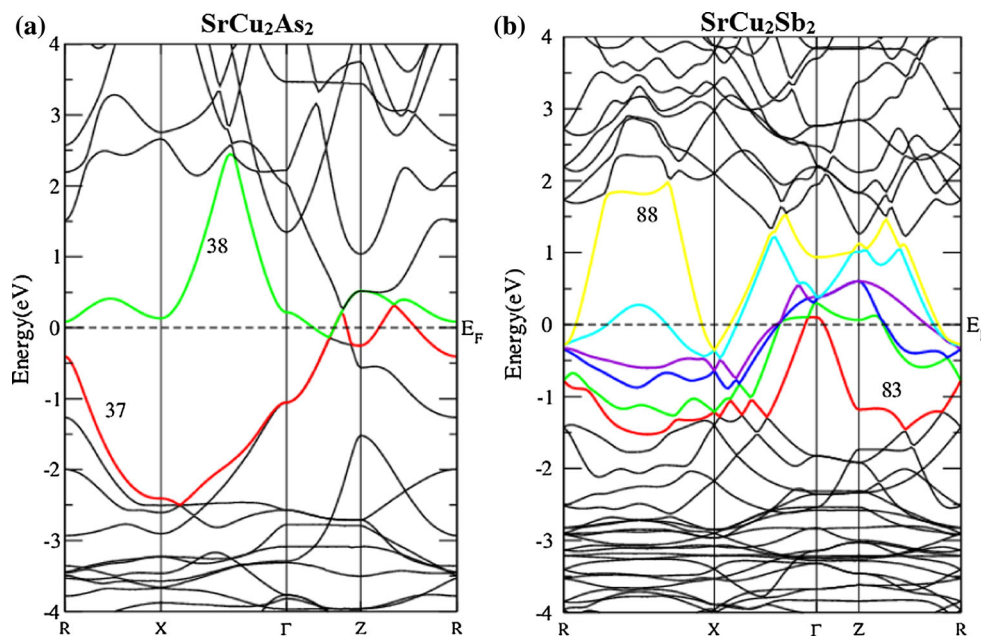


Fig. 2 Calculated band structure of: **a** SrCu_2As_2 , **b** SrCu_2Sb_2



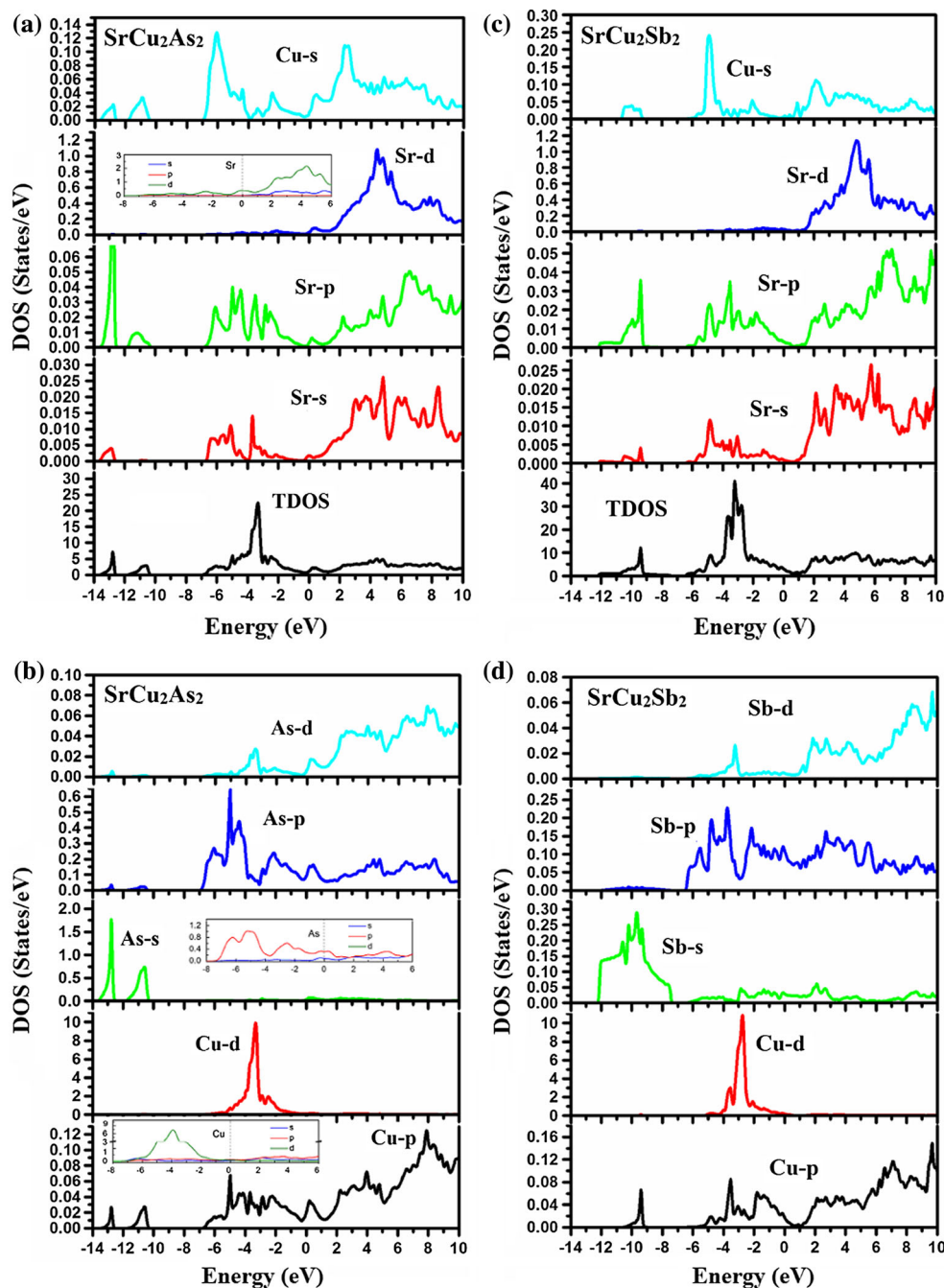
potential, which diverges as at the nucleus, as opposed to the pseudo-potential, in which the singularity is removed [28].

Crystal structure and computational details

In the present calculation, the crystallographic data of SrCu_2X_2 ($\text{X} = \text{As}, \text{Sb}$) are taken from Ref. [22]. The crystal structure of SrCu_2As_2 is stable in the body-centered tetragonal structure ($I4/mmm$), while SrCu_2Sb_2 shows the

stability in the primitive tetragonal structure ($P4/nmm$), as shown in Fig. 1a, b. The ground state calculations were carried out using full potential linear augmented plane wave (FP-LAPW) as implemented in WIEN2k package [29]. The Engel Voskov generalized gradient approximation (EVGGA) [29] was used for calculating the electronic structure and optical properties. Generally, EVGGA obtained by optimizing the exchange–correlation potential V_{xc} as an alternative of the corresponding energy E_{xc} [30]. This scheme yields better band splitting and structural properties that essentially depend on the correctness of the

Fig. 3 Calculated total and partial densities of states for: **a** SrCu_2As_2 , **b** SrCu_2Sb_2



exchange–correlation potential [31, 32]. During the calculation, we consider valence electrons corresponding to Sr $5s^2$, Cu $3d^{10}4s^1$, As $3d^{10}4s^24p^3$, and Sb $4d^{10}5s^25p^3$ electronic configurations.

The calculations are converged with minimum energy cutoff $R_{MT}K_{max}$ up to 7.0 corresponding to 1379 and 3245 plane waves for $SrCu_2As_2$ and $SrCu_2Sb_2$, respectively, where R_{MT} and K_{max} correspond to muffin-tin (MT) sphere radius and magnitude of the largest K vector in plane wave expansion. The selected R_{MT} is 2.0 atomic units (a.u.) for Sr, Cu, As, and Sb in both $SrCu_2As_2$ and $SrCu_2Sb_2$. The wave function inside the sphere was expanded up to $l_{max} = 10$, whereas the Fourier expansion of the charge density was $G_{max} = 12(a.u.)^{-1}$ for $SrCu_2As_2$ and $SrCu_2Sb_2$. The valence and core bands are separated by -6.0Ryd. The self-consistent calculations are converged within the difference in total energy of the crystal did not exceed $10^{-2}mRyd$ for succeeding steps. The self-consistent calculations were obtained by using 159 and 144 k points in the irreducible Brillouin zone (IBZ) for $SrCu_2As_2$ and $SrCu_2Sb_2$ compounds.

Result and discussion

Band structure and Fermi surface

The calculated electronic band structure of $SrCu_2X_2$ ($X = As, Sb$) compounds along the high symmetry points of BZ is shown in Fig. 2a, b. The high dispersion of calculated bands structures around the Fermi level shows high mobility of the charge carrier (heavy hole, light hole, electron) as compared to the low-lying bands. The valence and conduction bands cut the Fermi level confirming the metallic nature of both the compounds.

The calculated total density of states (TDOS) along with partial density of state (PDOS) for $SrCu_2X_2$ ($X = As, Sb$)

compounds is shown in Fig. 3a–d. Our calculated density of states of states of $SrCu_2As_2$ compound shows good agreement with the previous theoretical work [23]. In $SrCu_2As_2$ compound, we notice that at around -13.0 eV, Sr- s , Cu- s/p , and As- $s/p/d$ states are strongly contributed in bands formation. When we replaced As atom by Sb, one can see that the peaks of Sr- s/p and Cu- s/p states around -13.0 eV in $SrCu_2As_2$ compound are vanished, and a small Sr- s peak appeared between -11.0 and -9.0 eV in the PDOS of $SrCu_2Sb_2$ compound, while the small peak of Sr- p between -12.0 and -10.0 eV in $SrCu_2As_2$ becomes pronounced in $SrCu_2Sb_2$. In addition to that the whole structure of Sr- $s/p/d$ and Cu- s/p states in $SrCu_2Sb_2$ shifts toward higher energies by 1.0 eV with respect to that of $SrCu_2As_2$.

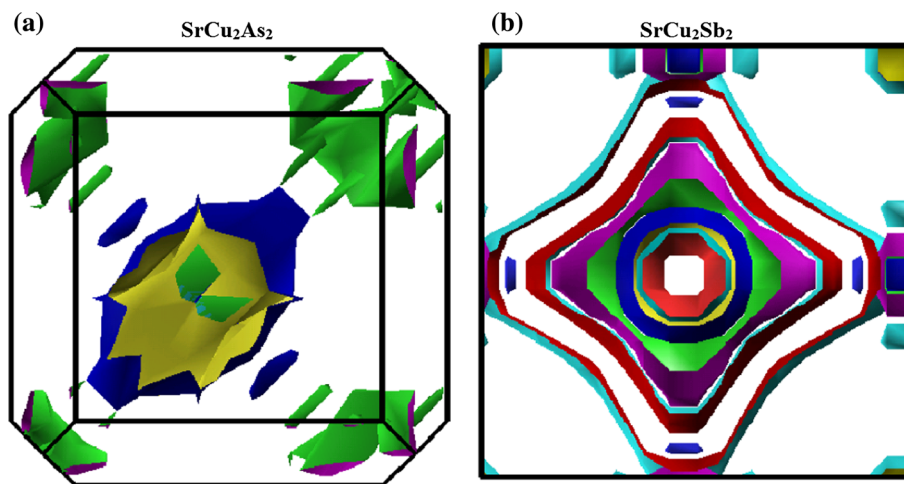
The density of states around Fermi level $[N(E_F)]$ in $SrCu_2As_2$ compound is formed by As- p/d , Sr- s , and Cu- s states while for $SrCu_2Sb_2$, compound is formed by Sb- p/d , Sr- s , and Cu- s states. The calculated values of $N(E_F)$ are 14.2 states/Ryd-cell and 42.57 states/Ryd-cell for $SrCu_2As_2$ and $SrCu_2Sb_2$ compounds, which show that the metallic nature of $SrCu_2Sb_2$ is three times more that that of $SrCu_2As_2$. The electronic specific heat (γ) is a function of $N(E_F)$ can be determined by using the expression [33]:

$$\gamma = \frac{1}{3} \pi^2 N(E_F) k_B^2, \tag{1}$$

where k_B represents Boltzman constant. The calculated values of γ are 2.60 mJ/mol K^2 (7.37 mJ/mol K^2) for $SrCu_2As_2$ ($SrCu_2Sb_2$) compounds, which confirm our previous observation that the metallic nature of $SrCu_2Sb_2$ is three times greater than that of $SrCu_2As_2$. The calculated value of γ for $SrCu_2As_2$ shows close agreement to the experimental value (2.22 mJ/mol K^2) [22].

Finally, the energy region extended from 1.0 to 10.0 eV, for $SrCu_2As_2$ compound is shaped by Sr- s/d , Cu- s/p , and As- $s/p/d$ states, whereas for $SrCu_2Sb_2$, it is formed by Sr- $s/p/d$, Cus/ p and Sb- $s/p/d$ states.

Fig. 4 Calculated Fermi surface for: **a** $SrCu_2As_2$, **b** $SrCu_2Sb_2$



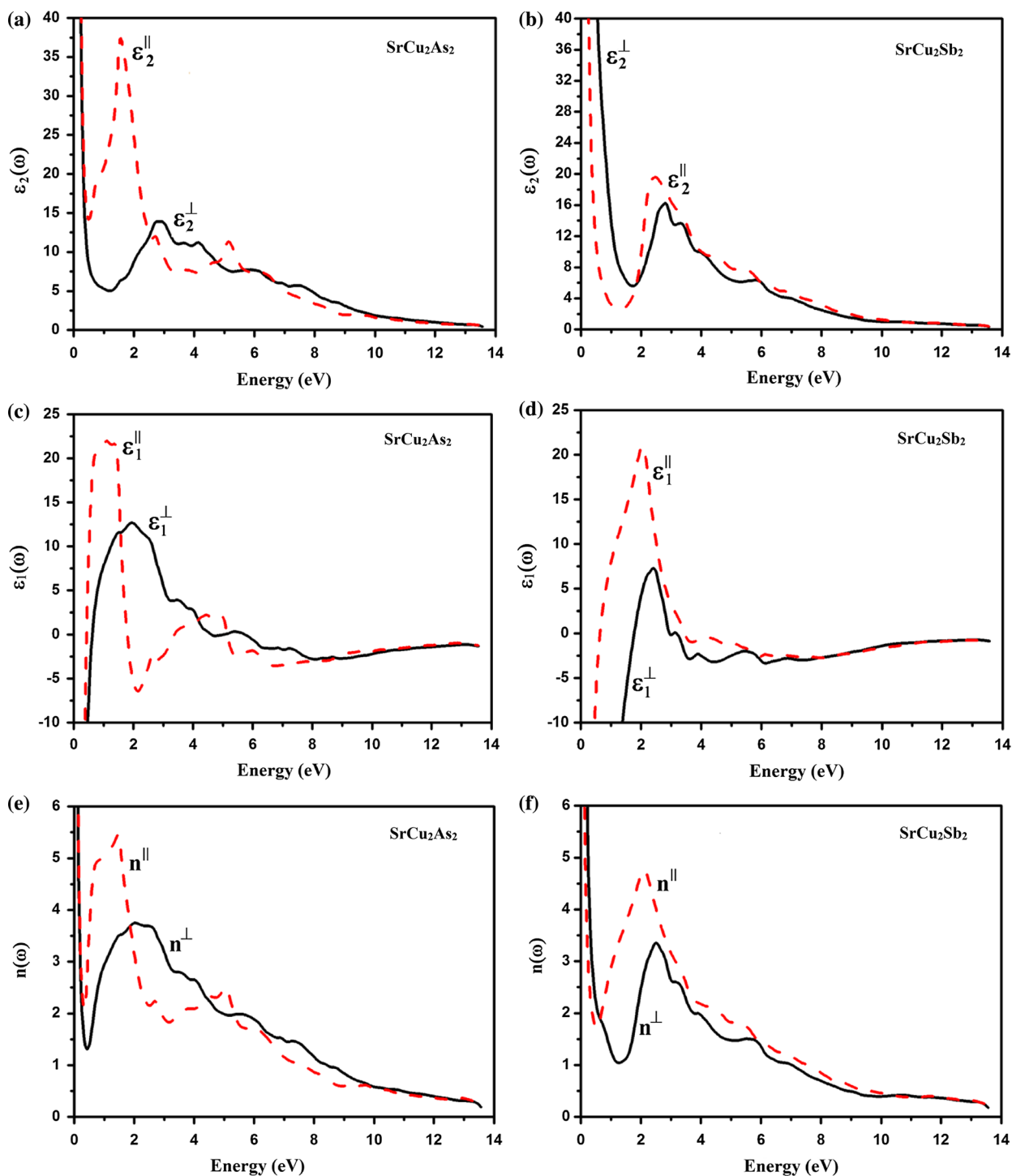


Fig. 5 **a** Calculated imaginary part of dielectric function for SrCu_2As_2 , **b** Calculated imaginary part of dielectric function for SrCu_2Sb_2 , **c** Calculated real part of dielectric function for SrCu_2As_2 , **d** Calculated real part of dielectric function for SrCu_2Sb_2 , **e** Calculated refractive index of SrCu_2As_2 , **f** Calculated refractive index of SrCu_2Sb_2 , **g** Calculated extension coefficient of SrCu_2As_2 , **h** Calculated extension coefficient of SrCu_2Sb_2 , **i** Calculated absorption coefficient of SrCu_2As_2 , **j** Calculated absorption coefficient of

SrCu_2Sb_2 , **k** Calculated reflectivity spectra of SrCu_2As_2 , **l** Calculated reflectivity spectra of SrCu_2Sb_2 , **m** Calculated energy loss function of SrCu_2As_2 , **n** Calculated energy loss function of SrCu_2Sb_2 , **o** Calculated real part of optical conductivity of SrCu_2As_2 , **p** Calculated real part of optical conductivity of SrCu_2Sb_2 , **q** Calculated imaginary part of optical conductivity of SrCu_2As_2 , **r** Calculated imaginary part of optical conductivity of SrCu_2Sb_2

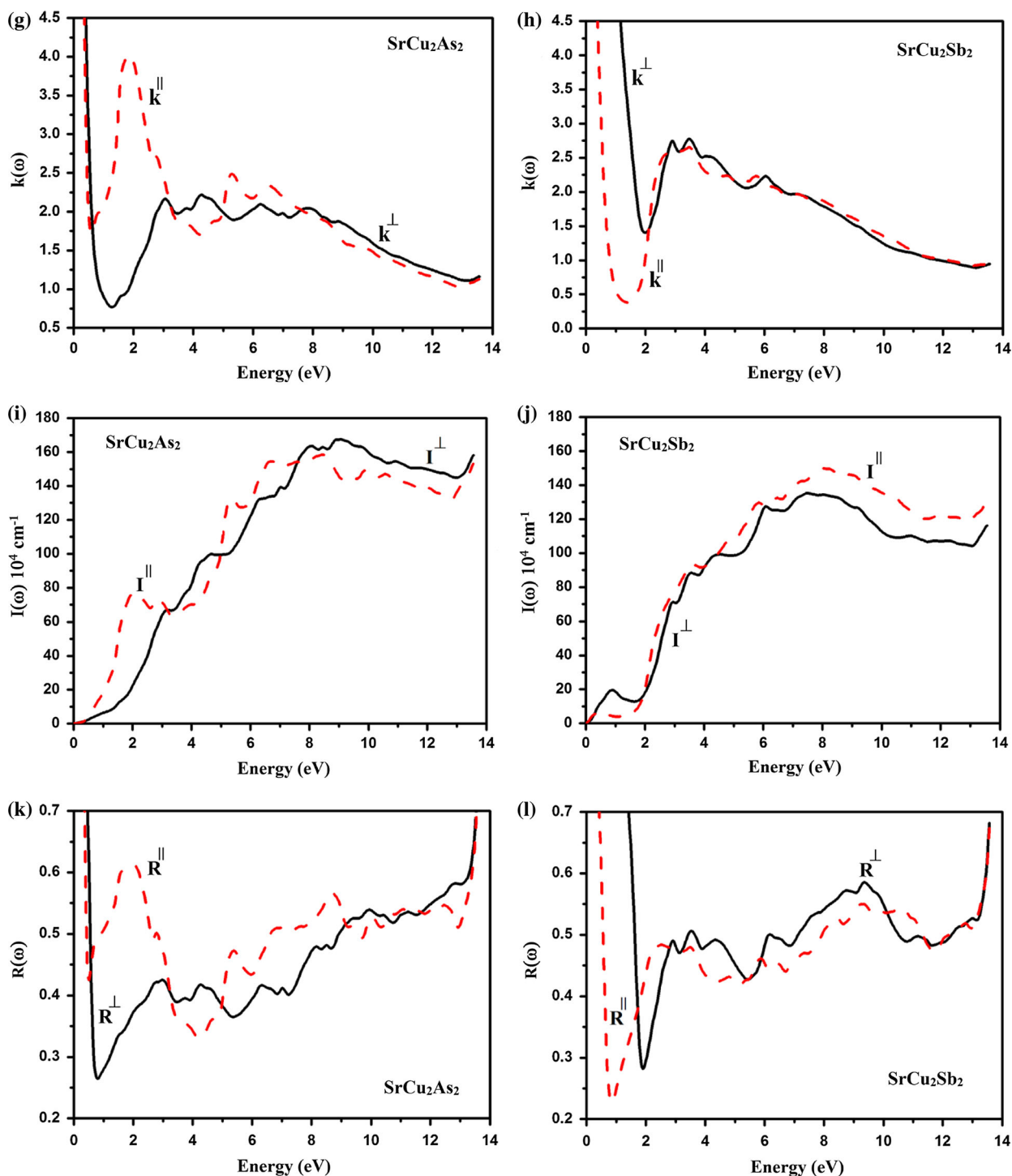


Fig. 5 continued

Figure 4a and b shows the Fermi surface of $\text{SrCu}_{-2}\text{X}_2$ ($\text{X} = \text{As}$ and Sb). For $\text{SrCu}_{-2}\text{As}_2$ compounds, the Fermi surface is formed by the bands #37 and 38, while it is formed from the bands #83, 84, 86, 87, and 88 in $\text{SrCu}_{-2}\text{Sb}_2$ compound. In the Fermi surface (FS), the white regions represent the hole concentration, while the colors show the

presence of electrons [34, 35]. The colors give an idea about the speed of the electrons at Fermi surface, the red color represents the highest speed electrons, the yellow, green, and blue colors exhibit the electrons with intermediate speed, whereas the violet color represents the electrons with the lowest speed [36]. Therefore, from Fig. 4a and b, and due to

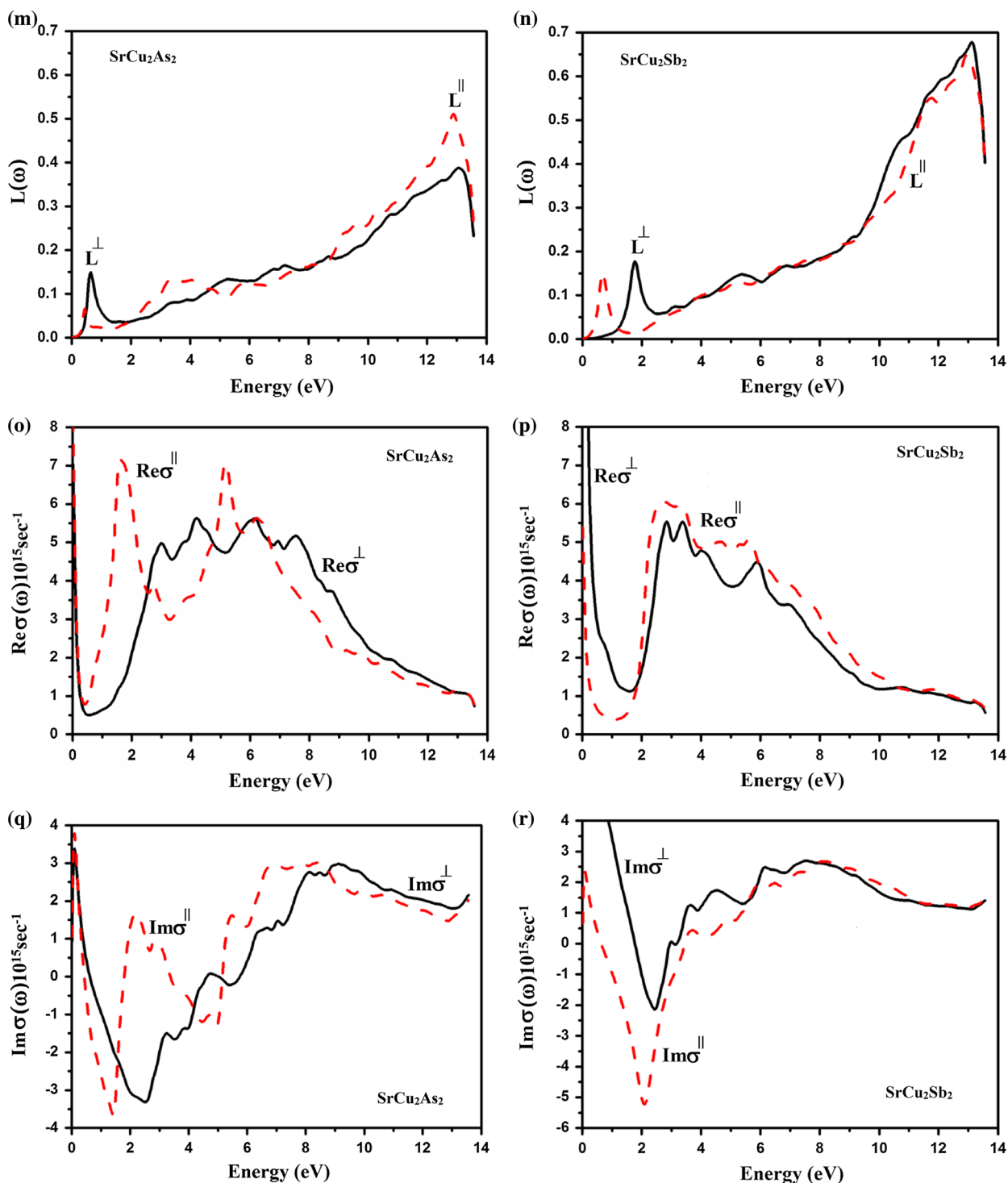


Fig. 5 continued

the colors of FS, one can notice that replacing As by Sb led to increase the metallic nature, which confirms our observation that the metallic nature of SrCu_2Sb_2 is greater than that of SrCu_2As_2 . In general, the transport properties are related to the electrons; these electrons are defined through Fermi surface, which determine the electrical conductivity.

Optical properties

The imaginary part of the frequency dependent dielectric function can be obtained from calculated band structure. The frequency-dependent dielectric function comprises both intra- and inter-band transitions. The intra-band

transitions, also known as inter-sub-band transitions, which occur between the quantized level in conduction or valence band are dominant in metals, whereas the inter-band transition occurs between the occupied and unoccupied bands. The correct energy eigenvalues and electron wave functions are necessary to calculate the frequency-dependent dielectric function $\epsilon(\omega)$. Since the crystal structures of SrCu_2As_2 and SrCu_2Sb_2 compounds are tetragonal, this symmetry allows only three nonzero dielectric tensor components. These are $\epsilon_2^{\text{xx}}(\omega) = \epsilon_2^{\text{yy}}(\omega)$ and $\epsilon_2^{\text{zz}}(\omega)$, for simplicity, we denote $\epsilon_2^{\text{xx}}(\omega)$ by $\epsilon_2^\perp(\omega)$ and $\epsilon_2^{\text{zz}}(\omega)$ by $\epsilon_2^\parallel(\omega)$. The imaginary part of the frequency-dependent dielectric function can be calculated using the expression given in Ref. 37:

$$\epsilon_2^{ij} = \frac{4\pi^2 e^2}{Vm^2\omega^2} \times \sum_{mm'\sigma} \langle kn\sigma | p_i | kn'\sigma \rangle \langle kn'\sigma | p_j | kn\sigma \rangle \times f_{kn}(1 - f_{kn'})\delta(E_{kn'} - E_{kn} - \hbar\omega), \tag{1}$$

where m and e stand for mass and charge of electron, the symbols ω and V represent the electromagnetic radiation strike the crystal, and unit cell volume, $|kn\sigma\rangle$, describes crystal wave function with crystal momentum k , and σ spin stands for the eigenvalue E_{kn} that corresponds to momentum operator p_j . The Fermi distribution function (f_{kn}) identifies the transition counting from occupied to unoccupied state, and $\delta(E_{kn'} - E_{kn} - \hbar\omega)$ shows the total energy conservation.

Since SrCu_2X_2 ($\text{X} = \text{As}, \text{Sb}$) compounds are metallic, therefore, we must include the Drude term (intra-band transitions) [38]:

$$\epsilon_2^\perp(\omega) = \epsilon_{2\text{intra}}^\perp(\omega) + \epsilon_{2\text{inter}}^\perp(\omega) \tag{3}$$

where

$$\epsilon_{2\text{intra}}^\perp(\omega) = \frac{\omega_p^{\perp 2}\tau}{\omega(1 + \omega^2\tau^2)}. \tag{4}$$

In the expression, τ represents the relaxation time, and ω_p^\perp stands for anisotropic plasma frequency [39]:

$$\omega_p^{\perp 2} = \frac{8\pi}{3} \sum_{kn} v_{kn}^{\perp 2} \delta(\epsilon_{kn}), \tag{4}$$

where v_{kn}^\perp shows electron velocity and ϵ_{kn} shows the difference between $E_n(k)$ and E_F . Similarly expressions for the parallel component can be written.

Figure 5a and b shows the calculated $\epsilon_2^\perp(\omega)$ and $\epsilon_2^\parallel(\omega)$ spectra of SrCu_2As_2 and SrCu_2Sb_2 . The sharp rise (<1.0 eV) in the optical spectral structure is due to Drude term [40]. The spectral structures in $\epsilon_2(\omega)$ (>1.0 eV) are caused by inter-band transitions. The first peak of SrCu_2As_2 (>1.0 eV) is formed by the transitions of electrons from bands #35, 36, 37 to band #38, 39, 40. The next peak is originated due to the transitions from band # 32, 33, 34 to

band #41, 42, 43 state. Whereas for SrCu_2Sb_2 , the first peak around 2.5 eV is due to the transitions from band #81, 82 to band #89, and the next peak around 3.4 eV is originated by the transitions from band #79, 80, 81 to band #90, 91, 92. There is a considerable anisotropy between $\epsilon_2^\perp(\omega)$ and $\epsilon_2^\parallel(\omega)$ for the entire spectral region.

The real part of the dielectric function can be obtained from imaginary part by means of Kramers–Kronig relation [41]:

$$\epsilon_1(\omega) = 1 + \frac{2}{\pi} P \int_0^\infty \frac{\omega' \epsilon_2(\omega')}{\omega'^2 - \omega^2} d\omega',$$

where P symbolizes the principal value of integral. Figure 5c shows $\epsilon_2^\perp(\omega)$ and $\epsilon_1^\parallel(\omega)$ of SrCu_2As_2 including the Drude term. There exists a considerable anisotropy between the two components from 0.6 to 8.0 eV, and then both components become isotropic. Following Fig. 5c, one can see that the first peaks of $\epsilon_1^\parallel(\omega)$ occur at 1.0 eV while that of $\epsilon_2^\perp(\omega)$ is situated at 2.0 eV. When we replace As by Sb, the first peak of $\epsilon_1^\parallel(\omega)$ shifts to be at 2.0 eV and for $\epsilon_2^\perp(\omega)$ at 2.5 eV (Fig. 5d). The two components $\epsilon_1^\parallel(\omega)$ and $\epsilon_2^\perp(\omega)$ show a considerable anisotropy extended from 0.5 to 7.0 eV.

The other optical constants such as refractive index $n(\omega)$, extension coefficient $k(\omega)$, absorption coefficient $I(\omega)$, reflectivity $R(\omega)$, energy loss function $L(\omega)$, and optical conductivity $\sigma(\omega)$ are shown in Fig. 5e–r. The calculated refractive indices of SrCu_2As_2 show considerable anisotropy between $n_2^\perp(\omega)$ and $n^\parallel(\omega)$ up to 9.5 eV as illustrated in Fig. 5e. The sharp rise below 1.0 eV is due to intra-band transitions. Then $n^\parallel(\omega)$ forms the first peak at 1.5 eV and $n_2^\perp(\omega)$ at 2.0 eV. As we move from As to Sb in the investigated compounds, we notice that the peak of $n^\parallel(\omega)$ shifts to be at 2.25 eV and for $n_2^\perp(\omega)$ to be at 2.5 eV with decreasing the peak heights (in Fig. 5f).

The extension coefficient (k) is more significant important phenomena in metals, which shows absorption of energy on surface of the material. The extension coefficient (k) of SrCu_2As_2 and SrCu_2Sb_2 compounds is presented in Fig. 5g and h. In SrCu_2As_2 and SrCu_2Sb_2 compounds, both of $k_2^\perp(\omega)$ and $k^\parallel(\omega)$ show sharp rise below 1.0 eV which is due to intra-band transitions. In SrCu_2As_2 compound At 2.0 eV, $k^\parallel(\omega)$ forms its first peak of about 3.99, while $k^\perp(\omega)$ shows broad structure extending between 3.0 and 8.0 eV. Replacing As by Sb causes significant changes in the spectral structure of $k^\perp(\omega)$ and $k^\parallel(\omega)$ below 2.0 eV that is attributed to the fact that in SrCu_2Sb_2 compound, more bands cut Fermi level, and the value of the density of states at Fermi level is three times greater than that of SrCu_2As_2 which makes SrCu_2Sb_2 compound more metallic.

The absorption spectra of SrCu₂As₂ and SrCu₂Sb₂ compounds are shown in Fig. 5i and j. For both compounds, the absorption coefficient increases with increasing the energy to reach the maximum values at 9.0 eV for SrCu₂As₂ and at 7.5 eV for SrCu₂Sb₂. Further increase of the energy causes to reduce the absorption coefficient.

The reflectivity spectra for both compounds are plotted in Fig. 5k and l; at low energy region, both compounds exhibit sharp rise which is due to intra-band transitions. For SrCu₂As₂, $R^{\parallel}(\omega)$ forms the first peak at around 2.0 eV, while for $R^{\perp}(\omega)$, the first peak occurs at 2.5 eV. The first valley is formed by $R^{\parallel}(\omega)$ at around 4.0 eV, whereas $R^{\perp}(\omega)$ forms its first valley at 5.5 eV. Then the reflectivity increases with increasing the energy. The reflectivity spectra of SrCu₂Sb₂, are illustrated in Fig. 5l; it shows that $R^{\parallel}(\omega)$ forms the first maximum at around 2.5 eV and at 3.5 eV for $R^{\perp}(\omega)$. The first valley occurs at 5.5 eV for both components.

$L(\omega)$ describes the energy loss of fast electron traveling in the material. The two components $L^{\perp}(\omega)$ and $L^{\parallel}(\omega)$ for SrCu₂As₂ and SrCu₂Sb₂ compounds are illustrated in Fig. 5m and n). The two components are increased with increasing the energy to reach the maximum value at 13.0 eV for both compounds. The sharp peaks produced in $L(\omega)$ are due to the plasma oscillation [42].

The calculated imaginary and real parts of optical conductivity assign response of the material to electromagnetic waves. The optical conductivity $\sigma(\omega)$ is correlated to the dielectric function $\varepsilon(\omega) = 1 + 4\pi i\sigma(\omega)/\omega$. The $\text{Re}\sigma^{\parallel}(\omega)$ spectra of SrCu₂As₂ (Fig. 5o) gain the maximum photocurrent at 1.8 eV, and $\text{Re}\sigma^{\perp}(\omega)$ shows maximum value at 4.3 eV. For SrCu₂Sb₂ (Fig. 5p), both components show maximum photocurrent at 2.8 eV. The imaginary part of optical conductivity of Fig. 5q and r) is related to the absorptive part of dielectric function (Fig. 5a, b). The $\text{Im}\sigma^{\parallel}(\omega)$ and $\text{Im}\sigma^{\perp}(\omega)$ vary in the same fashion as that of $\varepsilon_2^{\parallel}(\omega)$ and $\varepsilon_2^{\perp}(\omega)$. Both real and imaginary parts of optical conductivity show considerable anisotropy.

Conclusion

We have calculated the electronic band structure density of states, Fermi surface, and optical properties of SrCu₂X₂ (X = As, Sb) using all-electron-FPLAPW method within the framework of WIEN2 k. Engel Vosko approximation was used to treat the exchange correlation. The calculated electronic band structure of SrCu₂Sb₂ shows more bands cut Fermi level than that of SrCu₂As₂. The calculated values of $N(E_F)$ are 14.2 states/Ryd-cell and 42.57 states/Ryd-cell for SrCu₂As₂ and SrCu₂Sb₂ compounds, which show that the metallic nature of SrCu₂Sb₂ is three times more than that of

SrCu₂As₂. The calculated value of γ for SrCu₂As₂ shows close agreement to the experimental value (2.22 mJ/mol K²) [22]. The optical properties of SrCu₂As₂ and SrCu₂Sb₂ were calculated. Since these compounds are metallic, therefore, we have included the intra-band transition (Drude term). The spectral peaks of $\varepsilon_2(\omega)$ above 1.0 eV show electron transition from occupied to unoccupied states of the investigated compounds. The other optical constants such as refractive index, extension coefficient, absorption coefficient, reflectivity, energy loss function, and optical conductivity were calculated and discussed in detail. There exists a considerable anisotropy between the two components of all the optical properties of SrCu₂As₂ and SrCu₂Sb₂.

Acknowledgements The result was developed within the CENTEM project, reg. no. CZ.1.05/2.1.00/03.0088, co-funded by the ERDF as part of the Ministry of Education, Youth and Sports OP RDI programme.

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